The shapes of the curves have been analyzed by constructing difference tables, as described in appendix A. If we express the resistance as a power series function of the pressure $R = R_0 (1 + Ap + Bp^2 + Cp^3 ...)$, the coefficients A, B, C may be obtained from the difference tables. Only these three coefficients are needed to express our experimental results; they are included in table 4.

TABLE 3. A COMPARISON OF RESULTS FOR THE IDEAL RESISTIVITY OF POTASSIUM AT ZERO PRESSURE

$\rho_i/1^+$ (10 ° 12 cm deg K ⁻¹)			
(1)†	(2)†	(3)	(4)
2.360	2.360	2.3601	1.000
2.1337	· · · · · · · · · · · · · · · · · · ·	2.061	1.035
1.9796	· · · ·	1.872	1.058
1.919,		1.793	1.070
1.906		1.791	1.064
_ `	1.836,	1.780	1.031
	1.774	1.721	1.031
1.816.		1.689	1.075
1.798,		1.685	1.067
1.6495		1.538	1.073
1.662,		1.532	1.085
0.633		0.565	1.122
0.6386	0.619°	0.556	${1 \cdot 150 \\ 1 \cdot 113}$
0.5338		0.471	1.144
0.471_{2}		0.414	1.138
0.432		0.379	1.141
0.336_{4}	-	0.288	1.168
	$(1)^{\dagger}$ $2 \cdot 360_{1}$ $2 \cdot 133_{7}$ $1 \cdot 979_{6}$ $1 \cdot 919_{1}$ $1 \cdot 906_{4}$ $$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

 $\rho_i/T^* (10^{-8} \Omega \text{ cm deg K}^{-1})$

* Normalized to $2.3601 \times 10^{-8}\Omega$ cm deg K⁻¹ at 273.15 °K.

† Specimens in glass capillary tubes.

(1) Results from Woltjer & Kamerlingh Onnes (1924).

(2) Results from Meissner & Voigt (1930).

(3) This work.

(4) ρ_i (capillary specimen)/ ρ_i (bare wire).

3.1.3. The correction to constant density conditions

This correction is made in the way described in appendix A. The results of the calculations are given in tables 2 and 4 and they are also illustrated in figures 1 and 2. The systematic error given in table 2 arises from uncertainties in the equation of state of potassium; the error limits we have quoted are based on the supposition that at room temperature the error in the value of p', the pressure required to increase the density of potassium to its value at 0 °K under zero pressure, is 3 %.

3.2. Sodium

The results for sodium are similar in general form to those for potassium. Below about 40 °K there is, however, the extra complication of the martensitic transformation (cf. Dugdale & Gugan 1960). Details of the specimens studied are given in table 5.