The shapes of the curves have been analyzed by constructing difference tables, as described in appendix $A$. If we express the resistance as a power series function of the pressure $R=R_{0}\left(1+A p+B p^{2}+C p^{3} \ldots\right)$, the coefficients $A, B, C$ may be obtained from the difference tables. Only these three coefficients are needed to express our experimental results; they are included in table 4.

Table 3. A comparison of results for the ideal resistivity of potassium at zero pressure

| $T$ ( ${ }^{\circ} \mathrm{K}$ ) | $\rho_{i} / T^{*}\left(10^{-8} \Omega \mathrm{~cm} \mathrm{degK}{ }^{-1}\right)$ |  |  | (4) |
| :---: | :---: | :---: | :---: | :---: |
|  | (1) $\dagger$ | (2) $\dagger$ | (3) |  |
| 273.15 | $2 \cdot 360_{1}$ | $2 \cdot 360_{1}$ | $2 \cdot 360_{1}$ | 1.000 |
| $170 \cdot 9_{4}$ | $2 \cdot 133_{7}$ | - | 2.061 | 1.035 |
| 108.79 | $1.979_{6}$ | - | 1.872 | 1.058 |
| $90 \cdot 5$ | $1.919_{1}$ | - | 1.793 | 1.070 |
| $90 \cdot 21$ | $1.906_{4}$ | - | 1.791 | 1.064 |
| $87 \cdot 8$ | - | 1.836 ${ }_{2}$ | 1.780 | 1.031 |
| $77 \cdot 6$ | - | $1.774{ }_{6}$ | 1.721 | 1.031 |
| $72 \cdot 9$ | $1.816_{2}$ | - | 1-689 | 1.075 |
| $72 \cdot 20$ | $1.798_{3}$ | - | 1.685 | 1.067 |
| $56 \cdot 8$ | $1.649_{5}$ | - | 1.538 | 1.073 |
| 56.41 | $1 \cdot 662_{1}$ | - | 1.532 | 1.085 |
| $20 \cdot 62$ | $0.633_{9}$ | - | 0.565 | 1-122 |
| $20 \cdot 42$ | $0.638_{6}$ | 0.619 。 | $0 \cdot 556$ | $\left\{\begin{array}{l}1 \cdot 150 \\ 1 \cdot 113\end{array}\right.$ |
| 18.47 | $0.533_{8}$ | - | 0.471 | $1 \cdot 144$ |
| 17.19 | $0.471_{2}$ | - | $0 \cdot 414$ | 1.138 |
| 16.39 | $0.432{ }_{6}$ | - | $0 \cdot 379$ | $1 \cdot 141$ |
| 14.27 | $0.336_{4}$ | - | 0.288 | 1-168 |

* Normalized to $2.3601 \times 10^{-8} \Omega \mathrm{~cm}$ deg $\mathrm{K}^{-1}$ at $273.15^{\circ} \mathrm{K}$.
$\dagger$ Specimens in glass capillary tubes.
(1) Results from Woltjer \& Kamerlingh Onnes (1924).
(2) Results from Meissner \& Voigt (1930).
(3) This work.
(4) $\rho_{i}$ (capillary specimen) $/ \rho_{i}$ (bare wire).


## $3 \cdot 1 \cdot 3$. The correction to constant density conditions

This correction is made in the way described in appendix A. The results of the calculations are given in tables 2 and 4 and they are also illustrated in figures 1 and 2 . The systematic error given in table. 2 arises from uncertainties in the equation of state of potassium; the error limits we have quoted are based on the supposition that atroom temperature the error in the value of $p^{\prime}$, the pressure required to increase the density of potassium to its value at $0^{\circ} \mathrm{K}$ under zero pressure, is $3 \%$.

### 3.2. Sodium

The results for sodium are similar in general form to those for potassium. Below about $40^{\circ} \mathrm{K}$ there is, however, the extra complication of the martensitic transformation (cf. Dugdale \& Gugan 1960). Details of the specimens studied are given in table 5.

